

Errors and Uncertainty

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1.

A student carries out a series of determinations of the acceleration of free fall g . The table shows the results.

g/ms^{-2}
4.91
4.89
4.88
4.90
4.93
4.92

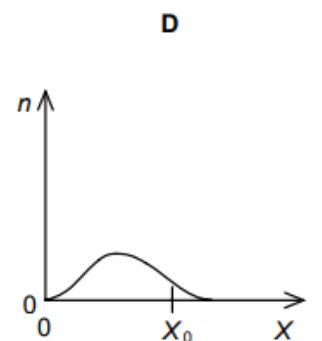
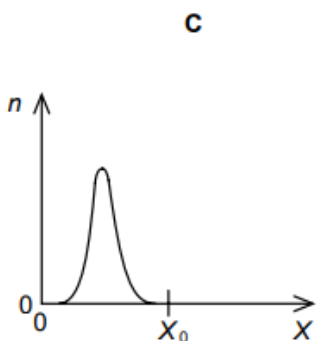
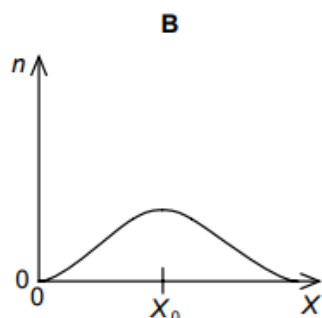
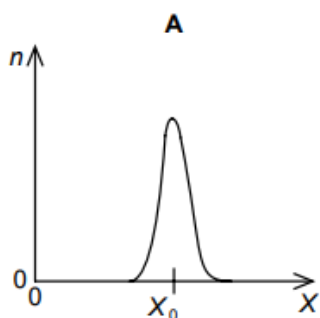
What can be said about this experiment?

- A** It is accurate and precise.
- B** It is accurate but not precise.
- C** It is not accurate and not precise.
- D** It is not accurate but is precise.

2.

A quantity X is measured many times. A graph is plotted showing the number n of times a particular value of X is obtained. X has a true value X_0 .

Which graph could be obtained if the measurement of X has a large systematic error but a small random error?



3.

A student makes measurements from which she calculates the speed of sound as 327.66 m s^{-1} . She estimates that her result is accurate to $\pm 3\%$.

Which of the following gives her result expressed to the appropriate number of significant figures?

- A** 327.7 m s^{-1} **B** 328 m s^{-1} **C** 330 m s^{-1} **D** 300 m s^{-1}

4.

A steel rule can be read to the nearest millimetre. It is used to measure the length of a bar whose true length is 895 mm. Repeated measurements give the following readings.

length / mm	892, 891, 892, 891, 891, 892
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Are the readings accurate and precise to within 1 mm?

	results are accurate to within 1 mm	results are precise to within 1 mm
A	no	no
B	no	yes
C	yes	no
D	yes	yes

5.

A student finds the density of a liquid by measuring its mass and its volume. The following is a summary of his measurements.

$$\text{mass of empty beaker} = (20 \pm 1)\text{g}$$

$$\text{mass of beaker + liquid} = (70 \pm 1)\text{g}$$

$$\text{volume of liquid} = (10.0 \pm 0.6)\text{cm}^3$$

He correctly calculates the density of the liquid as 5.0g cm^{-3} .

What is the uncertainty in this value?

- A** 0.3g cm^{-3} **B** 0.5g cm^{-3} **C** 0.6g cm^{-3} **D** 2.6g cm^{-3}

6.

A micrometer screw gauge is used to measure the diameter of a small uniform steel sphere. The micrometer reading is $5.00\text{ mm} \pm 0.01\text{ mm}$.

What will be the percentage uncertainty in a calculation of the volume of the sphere, using these values?

- A** 0.2% **B** 0.4% **C** 0.6% **D** 1.2%

7.

In a simple electrical circuit, the current in a resistor is measured as (2.50 ± 0.05) mA. The resistor is marked as having a value of $4.7 \Omega \pm 2\%$.

If these values were used to calculate the power dissipated in the resistor, what would be the percentage uncertainty in the value obtained?

- A** 2% **B** 4% **C** 6% **D** 8%

8.

Quantity X has a fractional uncertainty of x . Quantity Y has a fractional uncertainty of y .

What is the fractional uncertainty in $\frac{X}{Y^2}$?

- A** $x + y$ **B** $x - y$ **C** $x + 2y$ **D** $x - 2y$

9.

A person calculates the potential difference across a wire by using the measurements shown.

Which measured quantity has the greatest contribution to the percentage uncertainty in the calculated potential difference?

	quantity	value	uncertainty
A	current / A	5.0	± 0.5
B	diameter of wire / mm	0.8	± 0.1
C	length of wire / m	150	± 5
D	resistivity of metal in wire / Ω m	1.6×10^{-8}	$\pm 0.2 \times 10^{-8}$

10.

A digital meter has an accuracy of $\pm 1\%$.

The meter is used to measure the current in an electrical circuit.

The reading on the meter varies between 3.04 A and 3.08 A.

What is the value of the current, with its uncertainty?

A $(3.06 \pm 0.02) \text{ A}$

B $(3.06 \pm 0.04) \text{ A}$

C $(3.06 \pm 0.05) \text{ A}$

D $(3.06 \pm 0.07) \text{ A}$